

Title:**Curiosity at Gale crater, Mars: Characterization and analysis of the Rocknest sand
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Abstract:

The Rocknest aeolian deposit is similar to aeolian features analyzed by the Mars Exploration Rovers (MER) *Spirit* and *Opportunity*. The fraction of sand <150 μm in size contains ~55% crystalline material consistent with a basaltic heritage, and ~45% X-ray amorphous material. The amorphous component of Rocknest is Fe-rich and Si-poor, and is the host of the volatiles (H₂O, O₂, SO₂, CO₂, and Cl) detected by the Surface Analysis at Mars (SAM) instrument and of the fine-grained nanophase oxide (npOx) component first described from basaltic soils analyzed by MER. The similarity between soils and aeolian materials analyzed at Gusev crater, Meridiani Planum and Gale crater implies locally sourced, globally similar basaltic materials, or globally and regionally sourced basaltic components deposited locally at all three locations.

Introduction:

The Mars Science Laboratory rover *Curiosity* began exploring the surface of Mars on August 6, 2012, UTC; until September 13, 2012, it conducted an initial engineering checkout of its mobility system, arm and science instruments. *Curiosity* spent Sols 57-

100 (1) at a location named “Rocknest,” collecting and processing five scoops of loose, unconsolidated materials extracted from an aeolian sand shadow (2).

Five scoops of material from the Rocknest sand shadow were individually collected and sieved (<150 μm) by the Sample Acquisition, Sample Processing and Handling – Collection and Handling for In situ Martian Rock Analysis (SA/SPaH-CHIMRA) instrument (3). Scoops 1 and 2 were processed by CHIMRA and discarded to reduce (by entrainment and dilution) any terrestrial organic contamination that may have remained after a thorough cleaning on Earth (4) and to coat and “passivate” the interior surfaces of the collection device with Mars dust. Portions (40-50 mg.) of scoops 3 and 4 were delivered to the Chemistry and Mineralogy (CheMin) instrument (5) and the “observation tray,” a 7.5 cm diameter flat Ti-metal surface used for imaging and analyzing scooped and sieved material with *Curiosity*’s arm and mast instruments. Portions of scoop 5 were delivered to both CheMin and the SAM quadrupole mass spectrometer/gas chromatograph/tunable laser spectrometer suite of instruments (6).

We describe the physical sedimentology of Rocknest and suggest possible sources for the material making up the sand shadow. We utilize Alpha-Particle X-ray Spectrometer (APXS) and CheMin data to determine the amounts and chemistry of the crystalline and amorphous components of the sand shadow, and compare these results to global soil measurements from the MER rovers and to basaltic martian meteorites analyzed on Earth.

Results

Description and interpretation of the Rocknest sand shadow

The Rocknest sand shadow (7) is an accumulation of wind-blown sediment deposited in the lower-velocity lee of an obstacle in the path of the wind. The orientation of the sand shadow indicates that the constructive winds were from the north. The surface is composed of dust-coated, predominantly rounded, very coarse (1-2 mm) sand grains (Fig. 1a). Trenches created during the scooping show that these larger grains form an armored surface ~2-3 mm in thickness (Fig. 1b). Beneath the armored surface, the bedform interior consists of finer-grained material whose size distribution extends through the resolution limit of Mars Hand Lens Imager (MAHLI) images (~30 μm /pixel under the conditions of the observation) (8). Because of CHIMRA's 150 μm sieve, the larger grains that armor the surface could not be analyzed by CheMin.

Coarse sand grains that fell from the crust into the scoop-troughs lost their dust coating and show diversity in color, luster, and shape. Among the grains are gray and red lithic fragments, clear/translucent crystal fragments, and spheroids with glassy luster (Fig. 1c). Some grains showed bright glints in the martian sunlight, suggesting specular reflections from mineral crystal faces or cleavage surfaces (similar features were observed by the Optical Microscope onboard the Mars Phoenix Lander (9)). MAHLI images of a sieved portion of material deposited on the observation tray (3) showed a variety of particle types from clear to colored to dark, angular to spherical, and dull to glassy-lustered, (Fig. 1d).

During the scooping process, fragments of the armored surface were cohesive to the extent that “rafts” of surface crust were laterally compressed and displaced forward, and fragments of the crust fell into the scoop hole as cohesive units (Fig. 1b). The surface crust was also fractured and broken into ‘rafts’ during scuffing by rover wheels (a process by which an excavation is made into the subsurface of unconsolidated regolith by rotating a single rover wheel). Material beneath the crust also had some cohesion, as shown by the over-steep walls of the scoop scars (much greater than the angle of repose and vertical in some cases).

The sand shadow has a discernable internal structure. On the headwall and flanks of each scoop trench, a lighter-tone layer is apparent ~1 cm beneath and parallel to the dune surface (Fig. 1b). The origin of the layering is not understood, and three hypotheses are viable. First, the layering may represent changes in bulk composition or grain size that occurred during deposition. Second, the layering may be the result of changes in oxidation state or other chemical properties that occurred after deposition, in which case the conformable nature of the banding and the surface of the sand shadow reflect depth-dependent post-depositional chemical processes. Finally, the layering may represent zones richer or poorer in light-toned dust, reflecting times of lesser or greater sand accumulation relative to the air-fall dust.

The aeolian bedform at Rocknest is quite similar to coarse-grained ripples encountered at Gusev by the MER rover *Spirit* (10, 11) and at Meridiani Planum by MER rover *Opportunity* (12, 13) in that a coarse-grained, indurated, dust-coated surface overlies an

interior of markedly finer sediment. Coarse-grained ripples on Earth typically consist of a surface veneer of coarse grains and a finer grained interior (7, 14), and the martian bedforms have been considered analogous features (13, 15). The spatial grain-size sorting within coarse-grained ripples is thought to arise because of the short grain excursion length of the coarse grains traveling in creep and the much longer excursion length of finer saltating grains (16). With ripple migration, coarse grains are recycled through the bedform and become concentrated on the ripple surface where impacts from saltating grains tend to buoy the grains upward.

Although the dynamics of sand shadows differ from those of coarse-grained ripples, and sand shadows on Earth do not characteristically show a coarse-grained surface, similar dynamics may arise owing to the mix-load transport of grains in creep and saltation. Alternate interpretations are also possible. First, the coarse-grained surface could represent a lag formed as winds deflated finer grains. However, the paucity of coarse grains within the interior indicates that an unreasonable amount of deflation would have had to occur to produce the veneer. Second, the coarse-grained veneer could represent the terminal growth phase of the bedform. Because the size of a sand shadow is fixed by the upwind obstacle size (17), once the terminal size is approached, the lower wind speeds that characterize the wake and allow for deposition of finer sediment are replaced by wind speeds that approach the unmodified (primary) winds. At this point there would be selective deposition of coarse grains traveling in creep while finer saltating grains would bypass the bedform. Third, the sand shadow could have formed largely by the more readily transported fine saltation load, but as the area became depleted in finer grains,

more of the residuum of coarser grains would be incorporated into transport, with the coarse-grained surface arising through subsequent deflation.

None of these interpretations explains the general absence of observed coarse grains in the interior; the contrast in grain size between the surface and the interior is more marked in the Rocknest sand shadow and in some of the coarse-grained ripples observed by MER than in many Earth examples. This may reflect the greater impact energy of saltating grains on Mars compared to Earth, and their ability to transport disproportionately larger grains in creep (18). Regarding the apparent absence of interior coarse grains, the small scooped areas may not be representative of the entire bedform and interior horizons of coarse grains could easily have been bypassed. In addition, as seen with coarse-grained ripples on Earth, the amount of coarse sediment occurring in the interior varies and decreases with the supply coarse grains.

Regardless of the origin of the coarse-grained surface, this armored surface would stabilize the bedform during all but the strongest wind events. In turn, the armored surface would allow time for surface induration to develop, further stabilizing the sand shadow. The similarity of the armoring and induration of the sand shadow at Rocknest to coarse-grained ripples encountered by Spirit and Opportunity suggests that the processes of grain transport and stabilization are similar across equatorial Mars, and that Mars' winds (in recent eras) rarely were strong enough to transport sand grains of 1-3 mm diameter. To move the grains at the current atmospheric pressure of 0.02 kg/m^3 , the wind velocities would need to be $\sim 36 \text{ m/s}$ (80 mph) and $\sim 52 \text{ m/s}$ (116 mph) with and

without saltation, respectively. Under conditions of high obliquity, during which time the atmospheric pressure could increase to 0.04 kg/m^3 , these values would decrease to 26 m/s (58 mph) and $\sim 37 \text{ m/s}$ (83 mph) respectively (see Materials and Methods). The potential antiquity of the Rocknest sand shadow is highlighted by comparing it with granule ripples on Meridiani Planum, where cratering post-dates a field of pristine granule ripples and the crater count suggests an age of 50-200 ka (19).

Mineralogy of the Rocknest sand shadow

Analysis and interpretation of the mineralogy of the Rocknest sand shadow is given in a companion paper (20). Rocknest consists of both crystalline and X-ray amorphous components. The crystalline component is basaltic, comprised of plagioclase feldspar, forsteritic olivine, and the pyroxenes augite and pigeonite (20). All of the minor phases are consistent with a basaltic heritage with the exception of anhydrite and hematite. By constraining the compositions of the individual crystalline phases on the basis of their measured unit-cell parameters, the chemical compositions of the minerals of Rocknest were determined (21, 22).

The crystalline component of Rocknest is chemically and mineralogically similar to that inferred for martian basalts across the planet, and many of the basalts found in martian meteorites (Table 1) and, apart from somewhat lower Fe and K, broadly similar to estimates of the average Martian crust (23). These basalts all contain (or have chemical compositions consistent with) the minerals olivine, augite, pigeonite and plagioclase feldspar. The mineral proportions of the crystalline component of Rocknest are virtually

identical to those calculated for the unaltered Adirondack class basalts from Gusev Crater (CIPW normative mineralogy from their APXS analyses (Table 1) (24, 25). Chemically, the mafic minerals of the Rocknest sediment (olivine, augite, and pigeonite) are all consistent with high-temperature chemical equilibria among Ca, Fe, and Mg at $1050\pm75\text{C}$ (Fig. 2). This consistency with chemical equilibria suggests, but does not prove, that these minerals and the plagioclase feldspar all derived from a common basaltic source rock, which was broken down into individual grains or lithic fragments, and transported to Rocknest from regional source areas.

Bulk Chemistry of the Rocknest sand shadow

APXS provided an independent means of determining bulk chemistry of material in the Rocknest sand shadow. A measurement was made in a wheel scuff named Portage which was largely devoid of surface crust (Fig. 1a). The chemical composition (taking into account analytical uncertainty) is within 2σ of MER APXS analyses of basaltic soils (Table 2). The APXS chemistry of basaltic soils analyzed by the MER rovers at Gusev crater and Meridiani Planum landing sites (Table 2) are within 1σ of each other except for MgO and Na₂O, which are the same within 2σ (26-30). The MER compositional averages exclude soils that contain a substantial “local component” (high SO₃ and high SiO₂ for Gusev and high Fe₂O₃ for Meridiani). The near identity of compositions of the Rocknest, Gusev, and Meridiani basaltic soils implies either global-scale mixing of basaltic material or similar regional-scale basaltic source material or some combination thereof.

In contrast to the APXS measurement at the Portage wheel scuff, both CheMin and SAM measurements were carried out on the sieved, $<150\text{ }\mu\text{m}$ size fraction of soil. To discriminate potential differences between the fines delivered to CheMin and SAM and the bulk material analyzed in the wheel scuff, APXS chemistry was obtained from portions of sieved material deposited on the observation tray. APXS spectra from the bulk and sieved material are nearly identical, with the exception of a prominent Ti peak and increased background from the observation tray (reflecting Ti metal of the tray). Additionally, Ca, Mn and Fe signals in spectra from the observation tray are lowered proportionally as a function of their atomic number, suggesting that a fraction of these grains is smaller than the APXS sampling depth (31). Slightly elevated S and Cl, with a S/Cl ratio similar to that found in soils by MER (32), suggest a potential enrichment of these two elements in the $<150\text{ }\mu\text{m}$ fraction delivered to the observation tray.

In order to determine the amount and composition of the amorphous component, mass balance calculations were performed using the chemical composition of the bulk sample, the chemical compositions of the individual phases (plagioclase, sanidine, olivine, etc.) and the relative proportions of those phases in the crystalline component. The empirical formulas of the major crystalline phases (Table 3) and their chemical compositions (Table S2) were calculated from cell parameter data (20, 21, *Table S1*). The chemical formulas and compositions of the minor crystalline components were assigned by stoichiometry (e.g., ilmenite as TiFeO_3). The relative proportions of amorphous and crystalline components and their respective bulk compositions are summarized in Table 4, with Rocknest having ~45 wt.% amorphous and ~55 wt.% crystalline components

(33). The chemical compositions and proportions of amorphous and crystalline components were calculated on a light-element-free basis. The relative proportion of the amorphous component will in reality be greater than 45 wt.% because the volatile inventory is associated with that component (34).

Abundance estimates for the X-ray amorphous component of a sample may vary considerably depending on the method used for their determination. Bish et al. (20), for example, utilized a full pattern fitting method together with known amorphous standard materials analyzed in the laboratory to determine the amount of amorphous or poorly crystalline material contained in the CheMin X-ray diffraction pattern. Their reported value of ~27 wt. % \pm 50 % (1 σ range of 13-40 wt %), as calculated from diffraction and scattering data alone, is somewhat lower than the ~45% calculated from mass balance considerations, but both values are within the combined analytical uncertainty of the two techniques.

The inferred chemical composition of the amorphous component (Table 4) contains ~23% “FeO+Fe₂O₃”, suggesting that ferric nanophase oxide (“npOx” (25-26, 35)) is present in abundance. Similarly, S (principally contained within the amorphous component) is closely associated with the npOx in dunes at the MER sites (27, 28) as well. Abundances of SO₃ and Cl are correlated in soils from Gusev and Meridiani, which implies that both are associated with npOx in the amorphous component because these elements are not associated with Mg, Ca, or Fe in crystalline phases. The elements Cr, Mn, and P were associated with the amorphous component (Table 4), but they could

instead be present as crystalline phases (e.g. Ca-phosphate and chromite) at abundances below the CheMin detection limit and/or as substitutional impurities in the major crystalline phases (e.g., Mn and Cr in pyroxene).

The SAM instrument analyzed Rocknest for volatiles species and organic molecules (34), and it detected, in order of decreasing abundance: H₂O, SO₂, CO₂, and O₂. The crystalline phases, aside from a minor anhydrite component, do not include these species as a part of their structure, so they must either be present in the amorphous component or be present in the crystalline component at levels below the XRD detection limit, or both.

ChemCam spot observations in the scoop walls of Rocknest are characterized by the strong emissions from elemental hydrogen, although ChemCam is not sensitive to its bonding state (36). Comparison of this result with those of CheMin and SAM suggests that ChemCam detections of hydrogen most likely correspond to the H₂O associated with the amorphous component detected by CheMin.

Discussion

Global, regional and local sources

The crystalline phases in the Rocknest fines are consistent with a basaltic source and fit well within the measured qualitative mineralogy of basaltic martian meteorites and the normative mineralogy of Adirondack class olivine basalts at Gusev crater (25) (Table 1). If the Rocknest assemblage of basaltic crystalline and amorphous components is locally derived, it is distinct from mafic float rocks analyzed to date by APXS and ChemCam in Gale crater (36, 37). This observation suggests that the similarity in the chemical

compositions of aeolian bedforms (basaltic soil) at Gale, Gusev, and Meridiani (Table 2) might result from global-scale aeolian mixing of local-to-regional basaltic material that may or may not have variable chemical compositions. This process would require sufficiently strong winds occurring with sufficient frequency over a long enough time to achieve global or regional-scale transport of grains by saltation and suspension.

An alternative explanation for the comparable chemical compositions of aeolian bedforms at Gale, Gusev, and Meridiani is that the chemical compositions of martian basalts are similar at regional scales everywhere on the planet. The Rocknest sand shadow could reasonably have locally sourced 1-2 mm particles, with finer-grained regional basaltic material plus a contribution from global dust. The similarity of soil compositions (Table 2) suggests that the basaltic fine-grained materials at Gusev, Meridiani and Gale crater provide a reasonable approximation to the bulk composition of the exposed martian crust (38, 39).

It is tempting to suggest that the light-toned martian dust is largely represented by the Rocknest amorphous component. However, we have no data to show that the <150 μm size fraction (clay to fine-sand size fraction) of material analyzed by CheMin has its finest material preferentially enriched in amorphous material. The evidence from MER for basaltic soils suggests that the chemical composition of the fine-grained, light-toned soil is approximately the same as the coarser-grained, dark-toned soils (e.g., Table 10 of (40)).

The central mound of Gale crater (Mt. Sharp or Aeolis Mons) exhibits reflectance spectra suggesting the presence of crystalline hydrated sulfate minerals and

phylllosilicates (41), but neither was seen in Rocknest (above the 1-2% level). The absence of material from Mt. Sharp could arise from the wind pattern during formation of the Rocknest sand shadow – it is oriented so as to imply sediment transport from the north, and Mt. Sharp is east and southeast of Rocknest.

Materials and Methods

Calculation of windspeeds required to form the Rocknest sand shadow

The wind velocity required to move the coarse grains of the sand shadow by creep can be calculated. The critical shear velocity (u_{*c}) of the wind needed to transport 1 mm grains (d) is given by (44) as

$$u_{*c} = \sqrt{0.0123 \left(sgd + \frac{0.0003 \text{ kg/s}^2}{\rho_f d} \right)},$$

where $s = \rho_s / \rho_f$, ρ_s is the density of the grains using basalt (3000 kg/m^3), ρ_f is the density of martian air (0.02 kg/m^3), and g is the acceleration due to gravity (3.71 m/s^2). The calculated u_{*c} is 2.6 m/s , which represents the fluid shear velocity to initiate motion. Because grains in creep derive a portion of their momentum from collisions by saltating grains, on Earth once saltation begins creep can occur down to 0.7 u_{*c} (1.8 m/s as applied to the Rocknest grains), which represents the impact threshold for motion. Given a boundary layer created by winds blowing over the surface, shear velocities can then be related to the wind speeds above the surface by the law-of-the-wall

$$u_z = \frac{u_*}{k} \ln \left(\frac{z}{z_0} \right),$$

where u_z is the wind speed at height z above the surface (taken here as 1 m), k is a constant of 0.407, and z_0 is the roughness height where the idealized logarithmic wind

profile is predicted to be zero. Roughness height varies by grain size and the height of surface features such as wind ripples (7), and also by the height and intensity of the saltation cloud (45). Rocknest conditions are unknown, but z_0 is taken as 0.3 mm, which would be the roughness height with wind ripples 10 mm in height. Estimated wind speeds at 1 m above the surface are $\sim 52 \text{ m/s}$ (116 mph) and 36 m/s (80 mph) without and with saltation, respectively. As a result of the lower gravity and reduced atmospheric density on Mars, a greater hysteresis exists than on Earth between the fluid and impact thresholds, and saltation impacts upon grains are more energetic (18, 46-47). The combined effects suggest that initial transport of the coarse surface grains probably occurred at lower wind speeds than those calculated. Conversely, reactivation of the sand shadow would require significantly higher wind speeds because of induration of the surface.

Although observations from the Viking Lander 1 suggest that wind speeds of 30 m/s at a height of 1.6 m occurred during its two year lifetime (48), we do not know how often Mars winds can be capable of transporting 1-2 mm grains. The wind estimates above suggest that formation of the Rocknest sand shadow has involved rare strong winds, and that re-activation of the sand shadow from its currently indurated state would require even stronger and rarer winds.

Given the possibility of significant antiquity of the Rocknest sand shadow and similar coarse-grained bedforms on Mars, could their activation correspond to the martian obliquity cycle? At low obliquities the atmosphere collapses onto the polar caps, but at

high obliquity CO₂ is released to the atmosphere (49, 50). Taken as an end-member, atmospheric density may double at high obliquity and thereby enhance aeolian activity (50). As a comparison with the above values calculated for the present martian atmosphere, using 0.04 kg/m³ for atmospheric density, the calculated fluid u_{*c} is 1.9 m/s and the impact u_{*c} is 1.3 m/s, which correspond to wind speeds at the 1 m height of ~ 37 m/s (83 mph) and 26 m/s (58 mph), respectively. Although significantly lower than values calculated for present conditions, rare strong wind events are still implied.

References and Notes

- (1) A Mars solar day has a mean period of 24 hours, 39 minutes 35 seconds, and is customarily referred to as a “sol” to distinguish it from the roughly 3% shorter day on Earth.
- (2) A sand shadow is an accumulation of wind-blown sediment deposited in the lower-velocity lee of an obstacle in the path of the wind.
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(33) Because APXS does not discriminate among iron oxidation states, the total Fe concentration was proportioned in accordance with the oxidation state information carried by the crystalline phases (Table 3, column 3). FeO-Cryst and Fe₂O₃-Cryst are the concentrations of FeO and Fe₂O₃ required to accommodate olivine, augite, pigeonite, ilmenite, and magnetite and hematite, in accordance with their valence states. The remaining iron (FeO + Fe₂O₃) is then associated with the amorphous component without implications for oxidation state. Similarly, some SO₃ is reported as SO₃-Cryst to accommodate anhydrite as a crystalline component.

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Support from the NASA Mars Science Laboratory Mission is gratefully acknowledged. The chemical and mineralogical data presented here are derived from the archived data sets in the NASA Planetary Data System (PDS) <http://pds-geosciences.wustl.edu/missions/msl/>, specifically MSL-M-CHEMIN-2-EDR-V1.0 and MSL-M-APXS-2-EDR-V1.0. MBM was funded by the Danish Council for Independent Research/Natural Sciences (FNU grants 12-127126 and 11-107019). WG acknowledges partial funding by the Deutsche Forschungsgemeinschaft (DFG grant GO 2288/1-1). Some of this research was carried out at Jet Propulsion Laboratory, California Institute of Technology under a contract with the National Aeronautics and Space Administration.

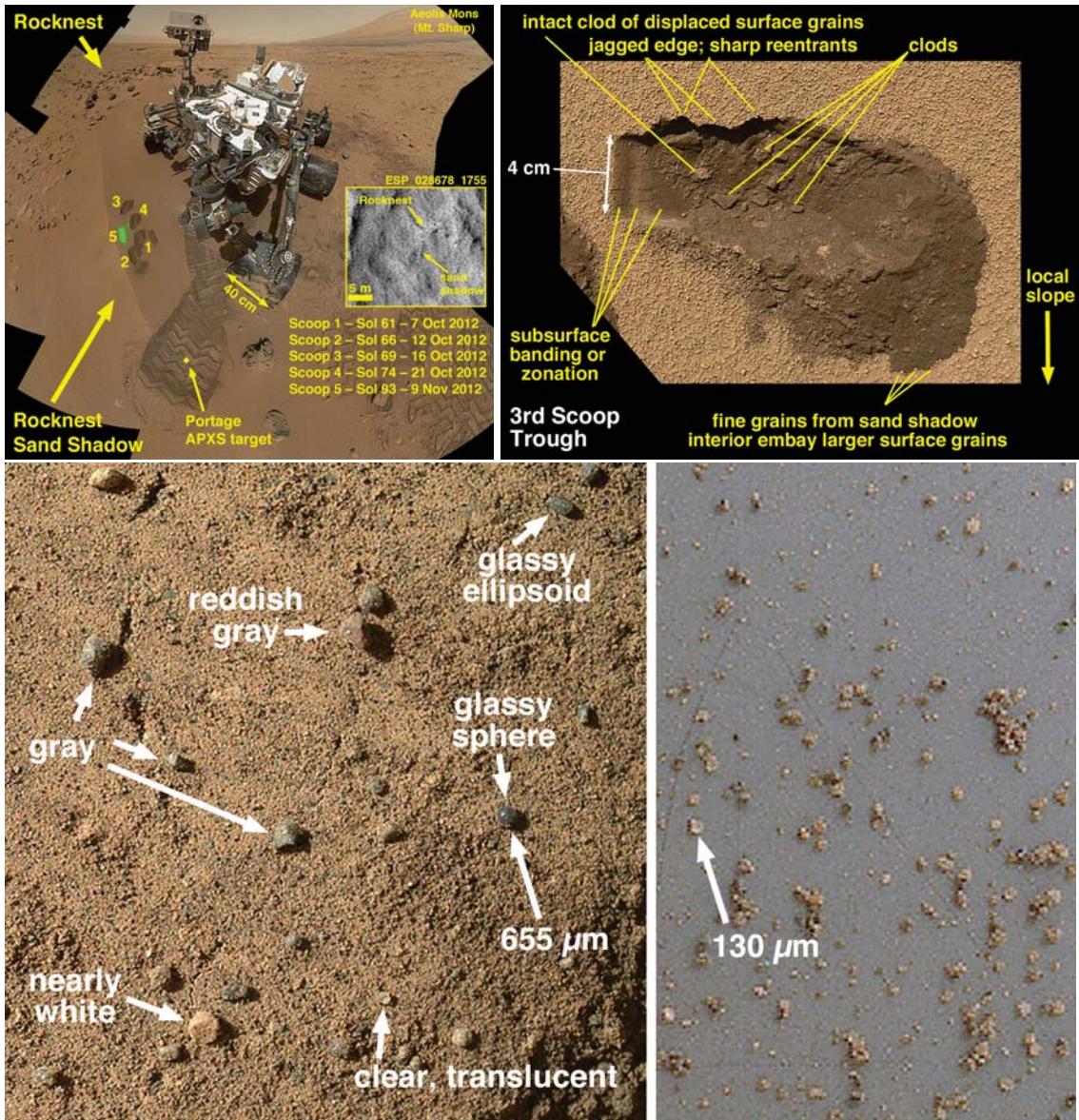


Figure 1. The Rocknest sand shadow, where *Curiosity* spent Sols 57-100 conducting engineering tests and science observations of the material. **(A)** (upper left) Mosaic of 55 MAHLI images showing Curiosity parked on the east side of the Rocknest sand shadow during the sampling campaign on Sol 84. The location of each of the five scoops is indicated. The inset is a portion of MRO HiRISE image ESP_028678_1755 showing the Rocknest sand shadow as seen from about 282 km above the ground. **(B)** (upper right) MAHLI image of third scoop trench, showing the dust-coated, indurated, armoring layer of coarse and very coarse sand and underlying, darker, fine and very fine sand. **(C)** (lower left) MAHLI image of Rocknest sand shadow surface disrupted by the rover's front, left wheel on Sol 57. The larger grains came from the armoring layer of coarse sand on the sand shadow surface. **(D)** (lower right), MAHLI image of a <150 μm sieved portion from the third scoop; grains similar to those delivered to the CheMin and SAM instruments, delivered to Curiosity's Ti observation tray.

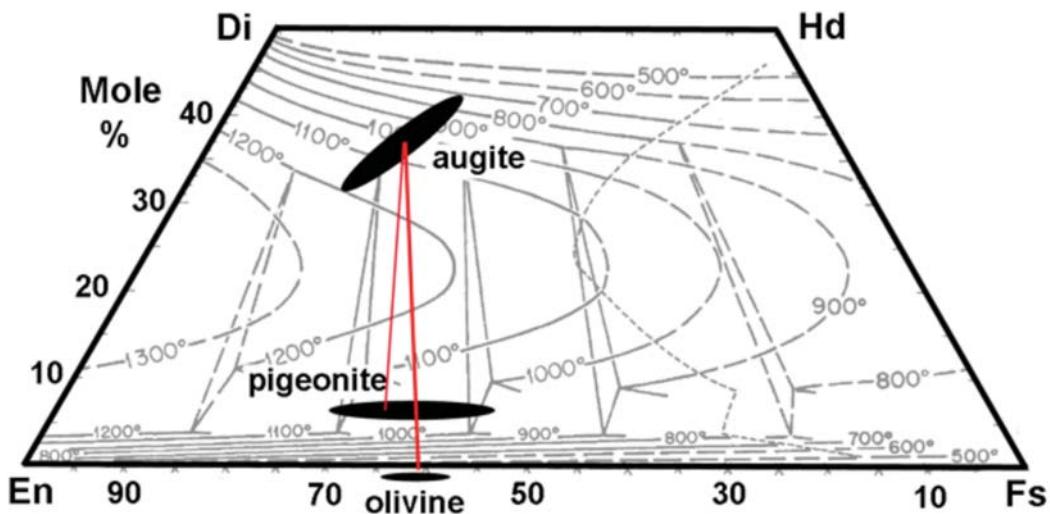


Figure 2. Pyroxene compositional quadrilateral, showing the chemical and thermal relationship between the major igneous minerals in the Rocknest sand shadow. Compositions of augite, pigeonite, and olivine in the Rocknest dune material, plotted on the pyroxene quadrilateral: En, enstatite, $\text{Mg}_2\text{Si}_2\text{O}_6$; Di, diopside, $\text{CaMgSi}_2\text{O}_6$; Hd, hedenbergite, $\text{CaFeSi}_2\text{O}_6$; and Fs, ferrosilite, $\text{Fe}_2\text{Si}_2\text{O}_6$. Pyroxenes are plotted within the quadrangle, based on CheMin XRD unit-cell parameters; olivine is plotted below the quadrilateral at the appropriate molar Mg/Fe ratio (20). Ellipses for each mineral approximate the uncertainties in mineral compositions from their unit-cell parameters. Gray background lines represent the surface of the pyroxene solvus, with temperatures in °C (42). Red lines are approximate equilibrium tie lines from the augite centroid composition to compositions of olivine and pigeonite, based on similar tie lines in an equilibrated anorthosite in lunar sample 62236 (43).

Table 1. Mineralogy of Rocknest soil (CheMin XRD) and normative mineralogies of basaltic materials from Gusev crater and of martian meteorites (Rocknest data are amorphous-free values).

Location	Gale	Gusev			Meteorites				
		Rocknest	Adirondack	Backstay	Irvine	Shergotty	NWA 6234		
Sample	Sand	Shadow					EETA 79001A		
Quartz	<i>1.4</i>		0	0	0	0.2	0	0	3
Plagioclase	<i>40.8</i>		39	49	32	23	19	19	32
K-spar	<i>1.3</i>		1	6	6	1	0.5	0	0
Low-Ca Pyx	<i>13.9</i>		15	14	21	46	30	47	15
High-Ca Pyx	<i>14.6</i>		15	5	13	25	16	16	38
Olivine	<i>22.4</i>		20	15	16	0	27	13	0
Fe-Cr oxides	<i>3.2</i>		6	4	6	3	4	2	0
Ilmenite	<i>0.9</i>		1	2	2	2	2	1	4
Apatite	-		1	3	2	2	2	1	6
Anhydrite	<i>1.5</i>								
Mg#	61±3		57	62	55	51	63	63	40
An	57±3		42	29	19	51	50	60	62

Notes for Table 1: Rocknest soil by CheMin (20), average of scoop 5, proportions of crystalline phases normalized to 100%; values in italics uncertain. CIPW norms (weight) for Gusev basaltic materials from MER APXS chemical analyses (26) ignoring S and Cl; $\text{Fe}^{3+}/\text{Fe}_{\text{tot}}$ for Backstay and Irvine taken as 0.17, the value for RATted Adirondack basalt (26). CIPW norms (weight %) of martian meteorites from bulk compositions [t-x]; $\text{Fe}^{3+}/\text{Fe}_{\text{tot}}$ as analyzed, and estimated at 0.1 for NWA 6234 and 0 for QUE94201. ‘K-spar’ is sanidine for the Rocknest soil, and normative orthoclase for others. ‘Low-Ca Pyx’ is pigeonite for the soil, and normative hypersthene for others. ‘High-Ca Pyx’ is augite for the soil, and normative diopside for others. ‘Fe-Cr oxide’ includes magnetite, hematite and chromite. All phosphorus in analyses calculated as normative apatite.

Table 2. Basaltic soil compositions from APXS analyses for Rocknest_Portage, Gusev Crater, and Meridiani Planum

	Rocknest	Gusev	Meridiani
Number	1 ^a	48 ^b	29 ^b
SiO ₂ (wt.%)	42.88 ± 0.47	46.1 ± 0.9	45.7 ± 1.3
TiO ₂	1.19 ± 0.03	0.88 ± 0.19	1.03 ± 0.12
Al ₂ O ₃	9.43 ± 0.14	10.19 ± 0.69	9.25 ± 0.50
Cr ₂ O ₃	0.49 ± 0.02	0.33 ± 0.07	0.41 ± 0.06
Fe ₂ O ₃ + FeO	19.19 ± 0.12	16.3 ± 1.1	18.8 ± 1.2
MnO	0.41 ± 0.01	0.32 ± 0.03	0.37 ± 0.02
MgO	8.69 ± 0.14	8.67 ± 0.60	7.38 ± 0.29
CaO	7.28 ± 0.07	6.3 ± 0.29	6.93 ± 0.32
Na ₂ O	2.72 ± 0.10	3.01 ± 0.30	2.21 ± 0.18
K ₂ O	0.49 ± 0.01	0.44 ± 0.07	0.48 ± 0.05
P ₂ O ₅	0.94 ± 0.03	0.91 ± 0.31	0.84 ± 0.06
SO ₃	5.45 ± 0.10	5.78 ± 1.25	5.83 ± 1.04
Cl	0.69 ± 0.02	0.7 ± 0.16	0.65 ± 0.09
Br (µg/g)	26 ± 6	53 ± 46	100 ± 111
Ni	446 ± 29	476 ± 142	457 ± 97
Zn	337 ± 17	270 ± 90	309 ± 87
Sum (wt.%)	99.85	99.88	99.88
Cl/SO ₃	0.13 ± 0.02	0.12 ± 0.02	0.11 ± 0.01

^aGellert et al., 2013 (37); analytical uncertainty.

^b1-σ standard deviation of average.

Table 3. Empirical chemical formulas of the four major phases identified in the Rocknest soil estimated by crystal-chemical techniques.

olivine	$(\text{Mg}_{0.62(3)}\text{Fe}_{0.38})_2\text{SiO}_4$
plagioclase	$(\text{Ca}_{0.57(13)}\text{Na}_{0.43})(\text{Al}_{1.57}\text{Si}_{2.43})\text{O}_8$
augite	$(\text{Ca}_{0.75(4)}\text{Mg}_{0.88(10)}\text{Fe}_{0.37})\text{Si}_2\text{O}_6$
pigeonite	$(\text{Mg}_{1.13(9)}\text{Fe}_{0.68(10)}\text{Ca}_{0.19})\text{Si}_2\text{O}_6$

Table 4. Chemical composition and proportion of XRD amorphous component in Rocknest Portage from APXS and CheMin data.

Origin	Remove XRD Crystalline Component							Composition					
	APXS ² CheMin	APXS ⁺ CheMin	Plagioclase	San-idine	Olivine	Augite	Pigeonite	Ilmenite	Hematite	Magnetite	Anhydrite	Quartz	Amorphous ³
SiO ₂ , wt. %	42.88	42.88	30.88	30.42	25.95	21.63	17.51	17.51	17.51	17.51	16.76	37.20	47.59
TiO ₂	1.19	1.19	1.19	1.19	1.19	1.19	0.93	0.93	0.93	0.93	0.93	2.06	0.47
Al ₂ O ₃	9.43	9.43	2.85	2.72	2.72	2.72	2.72	2.72	2.72	2.72	2.72	6.04	12.24
Cr ₂ O ₃	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	1.09	0.00
FeO+Fe ₂ O ₃ ⁴	19.19	10.43	10.43	10.43	10.43	10.43	10.43	10.43	10.43	10.43	10.43	23.14	-0.10
FeO-Cryst ⁵	---	7.37	7.37	7.37	3.31	2.29	0.59	0.35	0.35	0.00	0.00	-0.01	13.48
Fe ₂ O ₃ -Cryst ⁶	---	1.39	1.39	1.39	1.39	1.39	1.39	1.39	0.79	0.00	0.00	-0.01	2.55
MnO	0.41	0.41	0.41	0.41	0.41	0.41	0.41	0.41	0.41	0.41	0.41	0.91	0.00
MgO	8.69	8.69	8.69	8.69	4.97	3.72	2.19	2.19	2.19	2.19	2.19	4.86	11.86
CaO	7.28	7.28	4.65	4.65	4.65	3.19	2.87	2.87	2.87	2.87	2.53	5.61	8.67
Na ₂ O	2.72	2.72	1.62	1.62	1.60	1.60	1.60	1.60	1.60	1.60	1.60	3.56	2.03
K ₂ O	0.49	0.49	0.49	0.49	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.89	0.16
P ₂ O ₅	0.94	0.94	0.94	0.94	0.94	0.94	0.94	0.94	0.94	0.94	0.94	2.09	-0.01
SO ₃	5.45	4.96	4.96	4.96	4.96	4.96	4.96	4.96	4.96	4.96	4.96	11.01	-0.05
SO ₃ -Cryst ⁷	---	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.00	-0.01	0.90
Cl	0.61	0.61	0.61	0.61	0.61	0.61	0.61	0.61	0.61	0.61	0.61	1.35	-0.01
Sum	99.77	99.77	77.47	76.77	64.52	56.47	48.80	48.30	47.70	46.55	45.71	44.96	99.77
$\sum(\text{FeO}+\text{Fe}_2\text{O}_3)$	19.19	19.19	---	---	---	---	---	---	---	---	---	23.14	16.03
$\sum(\text{SO}_3)$	5.54	5.54	---	---	---	---	---	---	---	---	---	11.01	0.90
Relative to Whole Sample	22.3	0.7	12.3	8.0	7.6	0.5	0.6	1.2	0.8	0.8	0.8	45.3	54.7
Relative to XRD Crystalline	40.8	1.3	22.4	14.6	13.9	0.9	1.1	2.1	1.5	1.4	1.4	---	100.0

1. Plagioclase = An57; olivine = Fo62; Augite = En44Fs20W_{0.36} (Mg/Fe = 2.2 atomic); Pigeonite = En56Fs35Wo₈ (Fe/Mg = 1.6 atomic).

2. APXS chemistry from Gellert et al. (37).

3. Cr₂O₃ and MnO calculated with the amorphous component.

4. Total Fe as FeO+Fe₂O₃ because APXS does not distinguish oxidation states.

5. FeO required for Fe²⁺ crystalline phases (olivine, augite, pigeonite, ilmenite, and magnetite).

6. Fe₂O₃ required for Fe³⁺ crystalline phases (hematite and magnetite).

7. SO₃ required for crystalline SO₃ crystalline phase (anhydrite).

Supplementary Materials for

CURIOSITY AT GALE CRATER, MARS: CHARACTERIZATION AND ANALYSIS OF THE ROCKNEST SAND SHADOW

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This PDF file includes:

Supplementary Text
Figs. S1 to S4
Tables S1 and S2

Supplementary Text

Calculation of mineral compositions from their unit-cell parameters

The chemical compositions of the major mineral phases found in the Rocknest sediment fines were determined by relating their refined unit-cell parameters (Table S1, from (20)) to those published in the literature (22). The figures below show the unit-cell parameter versus chemical composition plots used to estimate the elemental compositions of olivine, plagioclase, augite and pigeonite (Table 3) and their empirical chemical formulas (Table S2).

Table S1.

Wt. % and refined unit-cell parameters of the major crystalline phases in Rocknest (20).

Mineral	Wt.%	2σ	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	α (°)	β (°)	γ (°)
plagioclase	40.8%	2.4%	8.177(6)	12.868(9)	7.113(5)	93.43(4)	116.26(2)	90.13(3)
forsterite	22.4%	1.9%	10.327(7)	6.034(7)	4.771(5)	90	90	90
augite	14.6%	2.8%	9.782(9)	8.939(9)	5.269(7)	90	106.25(9)	90
pigeonite	13.8%	2.8%	9.652(9)	8.92(1)	5.254(7)	90	108.0(1)	90

Olivine

The Mg-content of olivine was obtained from the variation of Mg-content versus the unit-cell volume using data from the fayalite-forsterite join (Fig. S1). The least-squares equation for the number of atoms of Mg per formula unit (Mg#) as a function of the unit-cell volume is:

$$\text{Mg\#} = -0.0578V + 17.801$$

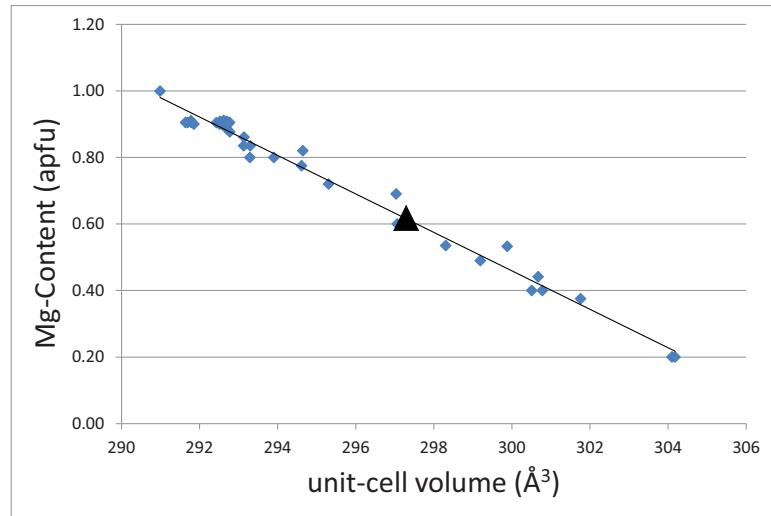


Fig. S1.

Mg-content of Fa-Fo olivine as a function of unit-cell volume. The black triangle represents the Rocknest olivine, indicating a composition of Fo 62.

Plagioclase

The refined unit-cell parameters provided information on the composition of plagioclase feldspar along the Ca-Na join. High-Ca feldspars are characterized by a c unit-cell parameter that is double that of lower-Ca feldspars; therefore, our data were restricted to the lower-Ca feldspars. Variations in Ca-content with unit-cell parameters are given in Fig. S2a, S2b and S2c. Fig. S2a shows that the b unit-cell parameter is not suitable for determining composition; therefore c and α were selected to determine the chemistry.

$$\text{Ca\#} = 157.0882779c^2 - 2251.2412721c + 8065.692 \text{ (Fig. S2b)}$$

$$\text{Ca\#} = 0.6896416\alpha^2 - 130.2226916\alpha + 6147.38751 \text{ (Fig. S2c)}$$

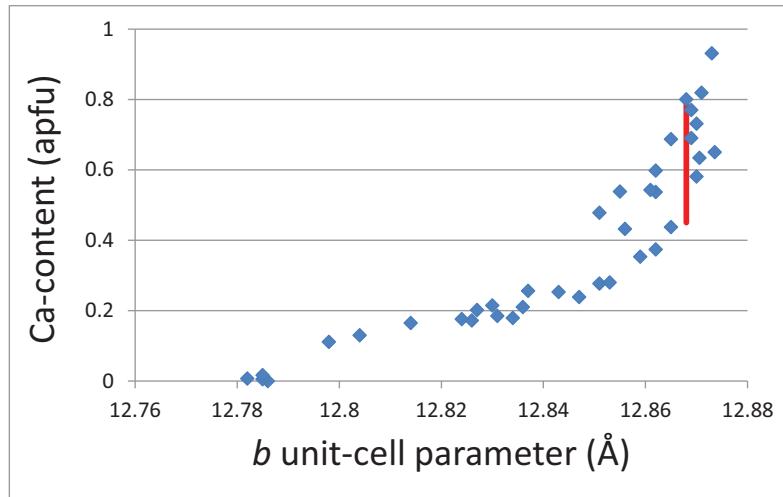


Fig. S2a

Ca-content of K-free plagioclase as a function of the b unit-cell parameter. The large scatter of data provides an error estimate of 12%.

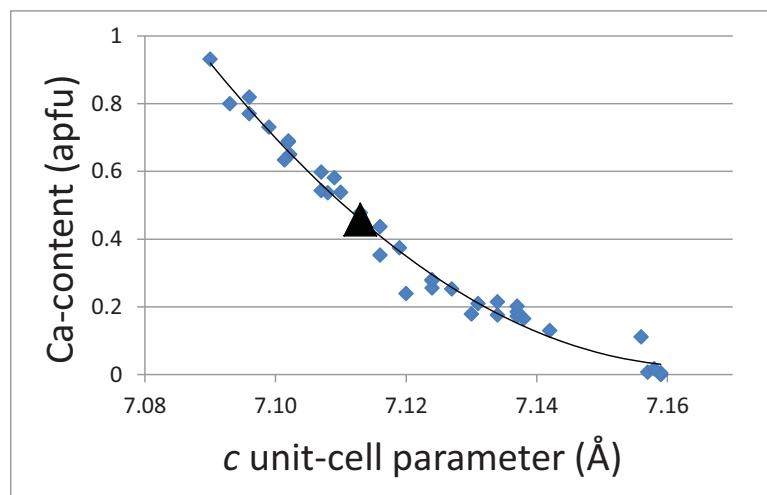


Fig. S2b

Ca-content of K-free plagioclase as a function of the c unit-cell parameter. This trend provides a lower bound on the amount of Ca (black triangle denotes Rocknest data).

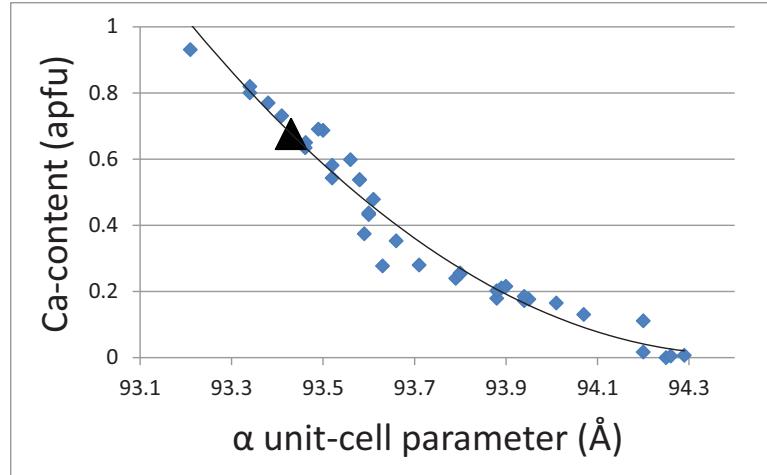


Fig. S2c

Ca-content of K-free plagioclase as a function of the α unit-cell parameter. This trend provides an upper bound on the amount of Ca (black triangle represents the Rocknest plagioclase).

Augite

Mg-content can be obtained from the variation of the b unit-cell parameter (Fig. S3a).

$$\text{Mg\#} = -9.8029b + 88.509$$

Fe/Mg content as a function of β has two sets of well-determined trends, that of $\text{Ca\#} = 1.0$ and $\text{Ca\#} = 0.8$ (Fig. S3b). Regressions of Mg-content for these two trends are:

$$\begin{aligned} \text{Mg\#} &= 0.9157\beta - 95.956 \quad (\text{Ca\#} = 1.0) \\ \text{Mg\#} &= 1.3719\beta - 144.74 \quad (\text{Ca\#} = 0.8) \end{aligned}$$

Substituting the value of Mg# determined from b produced β equal to 105.761° for the $\text{Ca\#} = 1.0$ trend and β equal to 106.145° for the $\text{Ca\#} = 0.8$ trend. Linear interpolation, using our value of β , gives $\text{Ca\#} = 0.75$. The value for $\text{Fe\#} = 2 - \text{Mg\#} - \text{Ca\#}$.

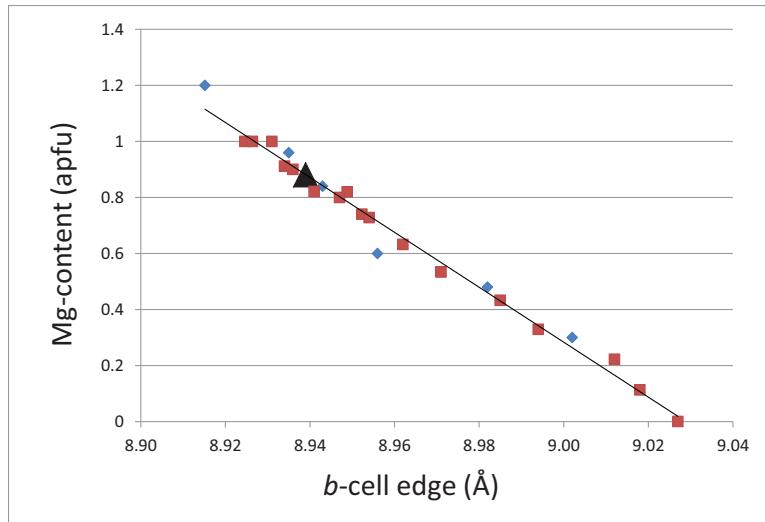


Fig. S3a

Variation of Mg-content with b unit-cell parameter in augite. Blue diamonds represent samples with Ca# = 0.8 and red squares are Ca# = 1. The black triangle represents the Rocknest augite – corresponding to a Mg# of 0.88.

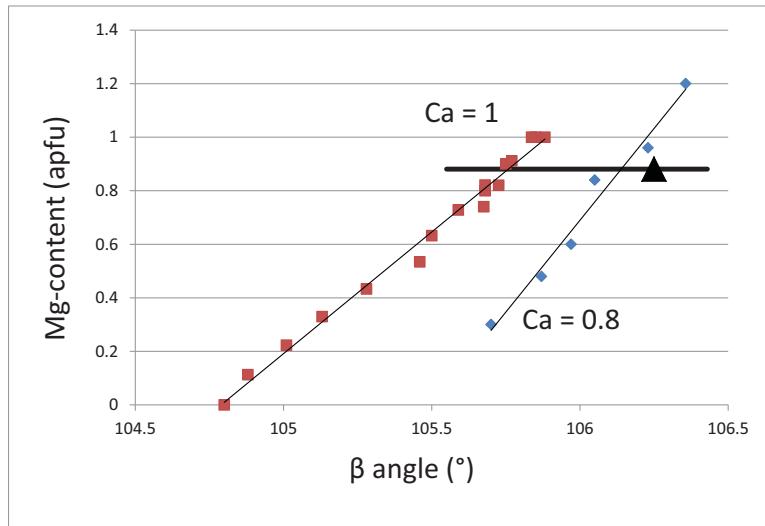


Fig. S3b

Ca-content in augite obtained by scaling the separation between the two trends of Mg-content versus β angle at Ca# = 1 (red squares) and Ca# = 0.8 (blue diamonds). The black triangle represents the Rocknest augite Mg-content at the refined β angle and corresponds to Ca# = 0.75.

Pigeonite

The Mg-content of pigeonite can be obtained from the b unit-cell parameter (Fig. S4a).

$$\text{Mg\#} = -8.9762b + 81.195$$

The Mg-Ca content as a function of unit-cell volume, V , has two sets of well-developed trends, that of $\text{Fe\#} = 1$ and $\text{Fe\#} = 0$ (Fig. S4b). Regressions of Mg-content for these two trends are:

$$\text{Mg\#} = -0.0471V + 21.148 \quad (\text{Fe\#} = 1)$$

$$\text{Mg\#} = -0.0254V + 12.572 \quad (\text{Fe\#} = 0)$$

Substituting the refined unit-cell volume from the Rocknest pigeonite gave $\text{Mg\#} = 0.89$ for the $\text{Fe\#} = 1$ trend and $\text{Mg\#} = 1.65$ for the $\text{Fe\#} = 0$ trend. Linear interpolation, using the Rocknest pigeonite Mg\# estimated from b (1.13), gives $\text{Fe\#} = 0.68$. The values for $\text{Ca\#} = 2 - \text{Mg\#} - \text{Fe\#}$.

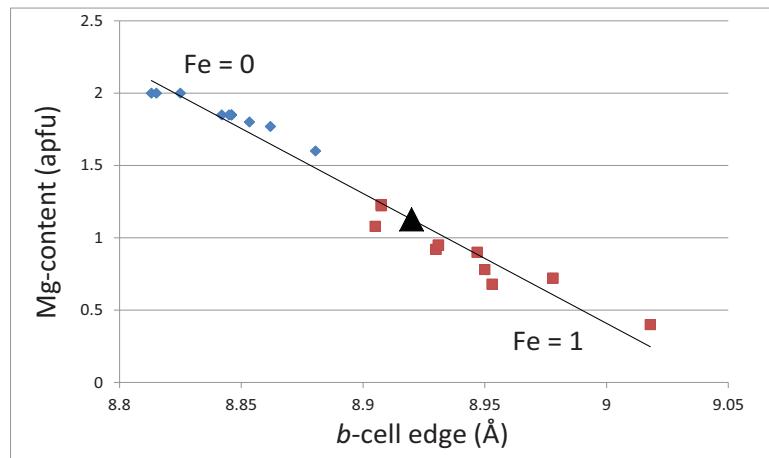


Fig. S4a

Variation of Mg-content with b unit-cell parameter in pigeonite. Blue diamonds represent $\text{Fe\#} = 0$ and red squares are $\text{Fe\#} = 1$. The estimated Mg\# of the Rocknest pigeonite is 1.13 and is marked with the black triangle.

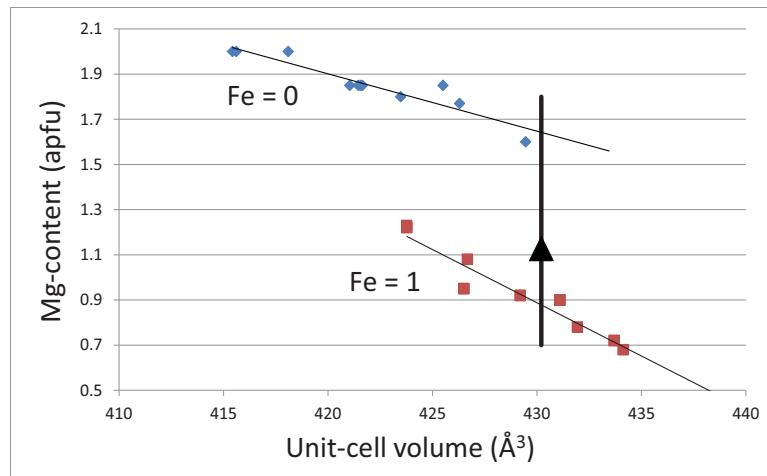


Fig. S4b

Fe-content in pigeonite - from the Mg-content determined above and scaling between the trends of samples with no Fe (blue diamonds) and Fe# = 1 (red squares), the Rocknest pigeonite is estimated to contain 0.68 apfu Fe.

Table S2.Oxide wt. % compositions of the crystalline mineral phases present in the <150 μm size fraction of Rocknest sediment

	augite	olivine	plagioclase	pigeonite	magnetite	hematite	ilmenite	sanidine	quartz	anhydrite	Total
SiO ₂	53.58	36.49	53.81	53.36	0	0	0	64.76	100	0	48.66
TiO ₂	0	0	0	0	0	0	52.65	0	0	0	0.37
Al ₂ O ₃	0	0	29.50	0	0	0	0	18.32	0	0	13.04
FeO	11.85	33.16	0	21.69	87	100	47.35	0	0	0	13.95
MnO	0	0	0	0	0	0	0	0	0	0	0.00
MgO	15.81	30.35	0	20.22	0	0	0	0	0	0	11.17
CaO	18.75	0	11.78	4.73	0	0	0	0	0	41.19	9.30
Na ₂ O	0	0	4.91	0	0	0	0	0	0	0	2.11
K ₂ O	0	0	0	0	0	0	16.92	0	0	0	0.36
SO ₃	0	0	0	0	0	0	0	0	58.81	0.82	
Component	0.146	0.224	0.408	0.138	0.021	0.009	0.013	0.014	0.015		

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